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DEVELOPMENT OF A MATHEMATICAL MODEL OF THE EFFECT OF LITHIUM ON THE WEAR RESISTANCE PROPERTIES OF ALUMINUM-LITHIUM ALLOYS

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ABSTRACT

In this research paper, the authors investigated the wear resistance of one of the mechanical properties, studying the change in its wear resistance as a result of the action of lithium fluoride on aluminum alloys. Based on the results obtained, a dependence graph was developed, on the basis of which a mathematical model of the effect of lithium on the edibility property was developed.

KEYWORDS

Aluminum, lithium, mathematical model, lithium fluoride, mechanical properties.

INTRODUCTION

Abrasive wear resistance is one of the most common types of wear resistance. This type of wear resistance

is found in the mining industry, agriculture, drilling equipment and tools, working bodies and chassis of

$$\begin{cases} 625\eta_1 + 25\eta_2 + 5\eta_3 = 0.04 \\ 1000\eta_1 + 1000\eta_2 + 10\eta_3 = 0.05 \\ 3375\eta_1 + 225\eta_2 + 15\eta_3 = 0.07 \end{cases} \quad (1)$$

Let's reduce the 1st system of equations to a matrix form and find the unknowns using Kramer's method to solve a complex system of linear equations. The determinant of the matrix, originally constructed using a system of equations, defines Δ .

$$\Delta = \begin{vmatrix} 625 & 25 & 5 \\ 1000 & 100 & 10 \\ 3375 & 225 & 15 \end{vmatrix} = -562500 \quad (2)$$

Solving the 2nd matrix, we define the following values of the unknowns,

$$\begin{aligned} x_1 &= \frac{\Delta_1}{\Delta} = \frac{-10}{-562500} = \frac{1}{56250} \\ x_2 &= \frac{\Delta_2}{\Delta} = -\frac{575/2}{562500} = -\frac{23}{45000} \\ x_3 &= \frac{\Delta_3}{\Delta} = \frac{-9375/2}{-562500} = \frac{1}{120} \end{aligned}$$

Having solved the above system and determined the unknowns, the function that expresses that the wear resistance of the alloy depends on the amount of lithium fluoride included in the alloy will look like this:

$$\varphi(\eta) = (1.78 \cdot 10^{-3}) \eta^3 - (5.1 \cdot 10^{-4}) \eta^2 + (8.3 \cdot 10^{-4}) \eta \quad (3)$$

In this order, we also calculate for D16 brand alloys.

$$\begin{cases} 625\eta_1 + 25\eta_2 + 5\eta_3 = 0.05 \\ 1000\eta_1 + 1000\eta_2 + 10\eta_3 = 0.06 \\ 3375\eta_1 + 225\eta_2 + 15\eta_3 = 0.075 \end{cases} \quad (4)$$

$$\Delta = \begin{vmatrix} 625 & 25 & 5 \\ 1000 & 100 & 10 \\ 3375 & 225 & 15 \end{vmatrix} = -562500 \quad (5)$$

$$\begin{aligned} x_1 &= \frac{\Delta_1}{\Delta} = \frac{-45/4}{-562500} = \frac{1}{50000} \\ x_2 &= \frac{\Delta_2}{\Delta} = -\frac{1575/4}{562500} = -\frac{7}{10000} \\ x_3 &= \frac{\Delta_3}{\Delta} = \frac{-12375/2}{-562500} = \frac{11}{1000} \end{aligned}$$

After solving the system in the same above order and identifying the unknowns, the function representing the dependence of the amount of lithium fluoride included in the alloy with liquid fluidity for the D16 brand alloy is as follows:

$$\varphi(\eta) = (2 \cdot 10^{-5}) \eta^3 - (7 \cdot 10^{-4}) \eta^2 + (11 \cdot 10^{-3}) \eta \quad (6)$$

CONCLUSION

It should be noted that from the derived functions, the degree of LaGrange polynomials can be generated by making them large enough based on data in tables and graphs. This is seen from the quotient value in front of the highest degree of the variable in polynomials, which allows us to obtain sufficient data in finite case.

Also, these functions make it possible to determine further results without experimentation, based on the results of the initial experiment.

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